Influence of different yeast/lactic acid bacteria combinations on the aromatic profile of red

Bordeaux wine

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1 ABSTRACT

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3 BACKGROUND: The typical fruity aroma of red Bordeaux wines depends on the grape variety but also on microbiological processes, such as alcoholic and malolactic fermentations. These 4 transformations involve respectively the yeast Saccharomyces cerevisiae and the lactic acid 5 bacterium Oenococcus oeni. Both species play a central role in red winemaking but their 6 7 quantitative and qualitative contribution to the revelation of the organoleptic qualities of wine has not yet been fully described. The aim of this study was to elucidate the influence of 8 9 sequential inoculation of different yeast and bacteria strains on the aromatic profile of red Bordeaux wine. 10

RESULTS: All microorganisms completed fermentations and no significant difference was observed between tanks regarding the main oenological parameters until 3 months' aging. Regardless of the yeast strain, B28 bacteria required the shortest period to completely degrade the malic acid, compared to the other strain. Quantification of 73 major components highlighted a specific volatile profile corresponding to each microorganism combination. However, the yeast strain appeared to have a predominant effect on aromatic compound levels, as well as on fruity aroma perception.

18 CONCLUSION: Yeasts had a greater impact on wine quality and have more influence on the19 aromatic style of red wine than bacteria.

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21 KEYWORDS

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23 Red wine, yeast, lactic acid bacteria, aromatic compounds

24 INTRODUCTION

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In the past, the aroma of red wines was characterized empirically by fruity notes and, 26 more specifically to Bordeaux wines, descriptors referred to red and black berry fruit, such as 27 raspberry, cherry, and blackcurrant. Recently, Pineau et al.(1) demonstrated the existence of a 28 sensory space specific to Bordeaux red wines. These fruity notes are not detected in must, but 29 are revealed during the various stages in winemaking and aging. Schematically, red 30 winemaking includes three important steps: alcoholic fermentation (AF), maceration and 31 malolactic fermentation (MLF). Fermentation processes play a central role in flavor 32 33 development and microorganisms, which take part in the vinification, act more or less in-depth on the composition of wine and through their action are largely responsible for its taste and its 34 aroma.(2) 35

36 During AF, yeasts such as Saccharomyces cerevisiae play a significant role in the formation and modulation of wine taste and aromas(3-5) by releasing varietal aromatic 37 compounds from grape precursors,(6,7) as well as synthesizing de novo volatile 38 compounds.(8,9) In contrast, the influence of MLF and lactic acid bacteria (LAB), such as O. 39 oeni, on red wine fruity aroma is not as clear. MLF is often empirically associated with a 40 decrease in the intensity of fruity notes. However, according to the literature, LAB enhance the 41 fruity aroma of red wines in some cases, attenuate it in others, and sometimes have no influence 42 on it at all.(10). These diverging results may be explained either by the use of different LAB 43 strains in these studies or by a matrix effect involving the cultivar and the yeast strain used to 44 carry out AF as well as the LAB. Indeed, it is well known that yeasts influence LAB growth 45 during winemaking.(11,12) Therefore, it would not be surprising that they also influence LAB 46 metabolism and thus the aromatic compounds in the wine. The few studies investigating these 47

effects demonstrated significant differences in the aroma of Chardonnay(13) and Chancellor
wines(14) fermented with several yeast/LAB strain combinations, at different temperatures.

The lack of fundamental data on the aromatic markers responsible for the fruity aroma 50 of red wines is probably another reason for the lack of consensus. Recent studies suggested that 51 these fruity notes were due to perceptive interactions between different families of aromatic 52 compounds, rather than individual compounds.(15,16) Varietal compounds, such as C13-53 54 norisoprenoids,(17) lactones,(18) thiols,(19) sulfur-containing compounds such as dimethyl sulfide,(20) and yeast- and LAB-derived compounds, including higher alcohols,(21) esters,(22) 55 volatile fatty acids,(23) and diacetyl(24) are examples of aromatic molecules that have a 56 57 negative or positive impact on red wine aroma.

This uncertainty surrounding the influence of fermentative microorganisms on wine 58 quality is problematic for winemakers. From a practical point of view, it would be useful to 59 60 know whether the influence of LAB strains on red wine quality is affected by some winemaking variables, particularly the yeast strain used for AF. Thus the aim of this study was to analyze 61 62 the impact of different yeast/LAB combinations on the pool of aromatic markers potentially responsible for the perception of fruity notes in red wines. Several combinations of yeast and 63 LAB were studied, using three commercial Saccharomyces cerevisiae strains and two 64 commercial O. oeni strains in sequential inoculation. Seventy-three compounds known to 65 contribute to the fruity notes of red wines were quantified using methods previously developed 66 in our laboratory. 67

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69 MATERIAL AND METHODS

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71 Yeast and Bacteria Strains

The three commercial Saccharomyces cerevisiae strains used in this work were Actiflore 72 cerevisiae® (522D), Zymaflore FX10® (Laffort, Floirac, France), and Excellence XR 73 (Lamothe-Abiet, Canéjan, France). Yeast implantation was verified by Polymerase Chain 74 Reaction (PCR) at the SARCO laboratory (Laffort, Floirac, France) (data not shown). Two 75 commercial O. oeni strains, Lactoenos 450 PreAc® and Lactoenos B28 PreAc® (Laffort, 76 Floirac, France), were used as MLF starters in this study. Bacteria implantation (data not 77 shown) was verified by the Microflora® laboratory (University of Bordeaux, France), using a 78 method developed by Claisse and Lonvaud-Funel.(25) 79

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81 Winemaking

Cabernet Sauvignon grapes from the Bordeaux appellation in the 2011 vintage were manually 82 harvested, destemmed, crushed and homogenously distributed into nine 2 hL stainless-steel 83 tanks (150 kg grapes per tank). Grape must was treated by adding pectolytic enzyme (Lafase® 84 Fruit, 0.03 µg g⁻¹, Laffort, Floirac, France) and yeast assimilable nitrogen was corrected to 85 around 210 mg N L⁻¹ by adding ammonium sulfate (Laffort, Floirac, France). Alcoholic 86 fermentation was conducted at 19-22 °C and initiated by inoculation with rehydrated dried 87 yeasts according to the manufacturer's recommendations. AF was performed in triplicate for 88 89 each yeast strain. Implantation in each tank was verified in the middle of AF (density close to 1.040). On completion of AF (<0.2 g L⁻¹ glucose/fructose), each 2 hL tank was divided into two 90 30 L stainless steel barrels for MLF. Bacterial cells were rehydrated with bacterial nutrient 91 (Energizer®, Laffort, Floirac, France) according to the manufacturer's instructions and 92 inoculated into wines at the recommended rate. For the entire duration of MLF, the malic acid 93 concentration was measured once per week to monitor the bacterial metabolism. At the end of 94 MLF (<0.1 g L⁻¹ malic acid), 50 g hL⁻¹ SO₂ was added. Wines were drained into 20 L stainless 95 steel barrels for 3 months' aging. After 3 months, wine composition was analyzed (total and 96

volatile acidity, total and free SO₂ content, pH, alcohol content) (Table 1). Samples were
collected for volatile compound analysis in 0.75 L glass bottles and stored at 10 °C for 1 week.
SO₂ content was measured and adjusted, if necessary. Wines were then decanted and frozen at
-18 °C until analysis.

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102 Standard Chemical Analyses

103 The standard chemical parameters of the wines (total acidity, sugar, malic acid, yeast 104 assimilable nitrogen, SO₂ content, pH, and alcohol) were analyzed by SARCO laboratory 105 (Laffort, Floirac, France), which has been accredited by COFRAC since 1995 (NF EN ISO 106 17025, accreditation N°1-0588). Analyses were carried out using the official methods or those 107 recommended by the International Organization of Viticulture and Wine (OIV)(26).

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109 Volatile Compound Analyses

Each wine sample was analyzed simultaneously after defrosting, which did not affect the aroma
compound concentrations in the racked wine. Eighty molecules were analyzed, using eight
different methods developed and validated in the laboratory.

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114 Chemicals

115 Commercial compounds were used as internal standards: butan-1,4-diol was obtained from 116 Merck (Damstadt, Germany); 4-methylpentan-2-ol (99%), octan-3-ol (99%), thiophene 117 (>99%), hexan-2,3-dione (97%), and ethyl-2-hydroxyisobutyrate (98%) were supplied by 118 Sigma-Aldrich (Steinheim, Germany), as well as 1,2-diaminobenzene (98%), used for 119 derivatization. Methanol (>99.9%), dichloromethane (>99%), phosphoric acid (85%), sodium 120 hydroxide (98%), sulfuric acid (98%) and sodium chloride (norma pure) were purchased from 121 VWR Chemicals (Fontenay-sous-Bois, France). Diethyl ether (>99%) and isohexane (>99%) were obtained from Carbo Erba Reactif-SDS (Val de Reuil, France) and ethanol (≥99.9%) from
Merck (Damstadt, Germany). Anhydrous sodium sulfate (99%) was supplied by Scharlau
Chemie (Sentmenat, Spain).

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126 Higher alcohols and ethyl acetate (direct injection and GC/FID analysis)

Propan-1-ol, 2-methylpropanol, 2-methylbutan-1-ol, 3-methylbutan-1-ol and ethyl acetate were 127 quantified using a modified version of the official OIV method (OIV-MA-AS315-02A). 128 According to this method, 5 mL wine was spiked with 50 µL internal standard solution (4-129 methylpentan-2-ol at 14.062 g L⁻¹ in 50% hydroalcoholic solution). The vials were filled with 130 131 this solution for direct injection into an HP 5890 gas chromatograph coupled to a flame ionization detector. The column was a CP-WAX 57 CB (50 m x 0.25 mm x 0.2 µm, Varian). 132 Quantification was performed using a calibration curve obtained from 12% hydroalcoholic 133 solution. 134

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136 Acetoin and butanediols (direct injection and GC/FID analysis)

The method developed by de Revel *et al.*(27) was used to quantify acetoin, D-butan-2,3-diol and meso-butan-2,3-diol. As specified in this method, 1 mL wine was spiked with 50 μ L internal standard solution (butan-1,4-diol at 1 g L⁻¹ in 40% hydroalcoholic solution) and diluted with 2 mL methanol. The vials were filled with this solution for direct injection into an Agilent 6890N gas chromatograph coupled to a flame ionizsation detector. The column was an FFAP type (BP21, 50 m x 0.25 mm x 0.2 μ m, SGE). Quantification was performed using a calibration curve obtained from 12% hydroalcoholic solution.

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145 Volatile fatty acids (liquid-liquid extraction and GC/FID analysis)

Butyric, hexanoic, octanoic, decanoic and dodecanoic acids were quantified using the method 146 developed by Bertrand.(28) In accordance with this method, 50 mL wine was spiked with 200 147 μ L internal standard solution (octan-3-ol at 400 mg L⁻¹ in 40% hydroalcoholic solution) and 0.3 148 mL phosphoric acid (diluted 1/3). Samples were successively extracted with 4 mL, 2 mL and 2 149 mL of a diethyl ether-isohexane mix (1:1, v/v). The organic phases were collected, dried with 150 anhydrous sodium sulfate and injected into an HP5890 gas chromatograph coupled to a flame 151 152 ionization detector. The column was an FFAP type (BP 21, 50 m x 0.25 mm x 0.2 µm, SGE). Quantification was performed with calibration curves obtained from red wines. 153

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155 Volatile sulfur compounds (headspace-gas chromatography-flame photometric detection 156 (HS-GC/FPD analysis))

Dimethyl sulfide (DMS) and hydrogen sulfide (H₂S) were quantified using the method developed and validated by Anocibar-Beloqui *et al.*(20) According to this method, 100 mL wine was spiked with 10 μ L internal standard solution (thiophene at 300 mg L⁻¹ in ethanol) in a 125 mL headspace vial. After 24 h at 22 °C, 1 mL of the gas phase was taken from the headspace and injected into an HP5890 gas chromatograph coupled to a flame photometric detector. The column was an HP5 (30 m x 0.25 mm x 0.25 μ m, Agilent). Quantification was performed using a calibration curve obtained from 12% hydroalcoholic solution.

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165 Diacetyl (liquid-liquid extraction after derivatization and gas chromatography-mass 166 spectrometry (GC-MS) analysis)

167 The method developed by de Revel *et al.*(27) was used for direct quantification of diacetyl, 168 glyoxal, methylglyoxal and pentan-2,3-dione. In accordance with this method, 50 mL wine was 169 spiked with 100 μ L internal standard solution (hexan-2,3-dione at 3.80 g L⁻¹ in 50% 170 hydroalcoholic solution). Then, 5 mL 1,2-diaminobenzene was added and pH was adjusted to 171 8 with NaOH (10 mol L⁻¹). After a 3 h derivatization reaction at 60 °C, the pH of the mixture 172 was adjusted to 2 with sulfuric acid (2 mol L⁻¹) and it was extracted twice with 5 mL 173 dichloromethane. The organic phases were collected, dried with anhydrous sodium sulfate and 174 injected into an Agilent 6890N gas chromatograph coupled to a mass spectrometer (Agilent 175 5973). GC-MS analysis conditions were as previously described.(27) Quantification was 176 performed with a calibration curve obtained from 12% hydroalcoholic solution.

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178 C₁₃-Norisoprenoids and lactones (stir bar sorptive extraction GC/MS)

This method, developed and validated by Antalick et al.,(29) was used to quantify four C13-179 180 norisoprenoids (β -damascenone, β -damascone, β -ionone, and α -ionone) and six lactones (γ octalactone, γ -nonalactone, γ -decalactone, γ -undecalactone, γ -dodecalactone, and δ -181 decalactone). According to the method, 25 mL wine was spiked with 25 µL internal standard 182 solution (ethyl-d5 cinnamate at 1.74 g L⁻¹ in ethanol) and a 20 mL sample was introduced into 183 a 25 mL vial. A 20 mm × 1 mm (length × film thickness) PDMS stir bar (Twister®, 126 µL 184 coating) (Gerstel, Müllheim an der Ruhr, Germany) was dropped into the vial, which was 185 capped with a PTFE-faced rubber stopper. The closed vial was stirred at 900 rpm for 1 h at 186 room temperature. At the end of the extraction time, the Twister® was removed from the vial, 187 washed quickly with Milli-Q water, and dried with lint-free tissue. Each Twister® was then 188 transferred into a glass tube for thermal desorption (Gerstel) and GC-MS analysis, under the 189 conditions described previously.(29) Quantification was performed using calibration curves 190 obtained from red wines. Ethyl-d5 cinnamate was synthesized using the method described by 191 Antalick *et al.*(30) 192

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194 Apolar esters (HS-SPME-GC-MS)

The method developed and validated by Antalick et al.(30) was used to quantify 32 esters: six 195 196 ethyl fatty acid esters, seven higher alcohol acetates, four ethyl branched acid esters, four methyl esters, three isoamyl esters, three ethyl esters with odd numbers of carbon atoms, two 197 ethyl cinnamates and some other minor esters. A mixture of ethyl-d5 butyrate, ethyl-d5 198 hexanoate, ethyl-ds octanoate and ethyl-ds cinnamate at about 200 mg L⁻¹ in ethanol was used 199 as internal standard. Deuterated esters were synthesized as described by Antalick *et al.*(30) In 200 201 accordance with this method, 20 µL internal standard solution was added to 25 mL wine. An aliquot of 10 mL was introduced into a 20 mL standard headspace vial containing 3.5 g sodium 202 chloride. The samples were extracted by HS-SPME and analyzed by GC-MS. The fiber used 203 204 was 100 µm polydimethylsiloxane (PDMS-100) (Supelco, Bellefonte, PA, USA), conditioned before use as recommended by the manufacturer. Quantification was performed with calibration 205 curves obtained from red wines. 206

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208 Additional volatile compounds (liquid-liquid extraction and GC/MS analysis)

209 The method developed and validated by Antalick et al.(29) was used to quantify seven polar esters (ethyl lactate, ethyl leucate, ethyl succinates and hydroxylated ethyl esters), three 210 branched acids (isobutyric acid, isovaleric acid and 2-methylbutyric acid), frambinone and 211 linalol. According to this method, 50 mL wine was spiked with 10 µL internal standard solution 212 (ethyl-2-hydroxyisobutyrate at 0.96 g L⁻¹ in ethanol). The mixture was successively extracted 213 with 4 mL, 2 mL, and 2 mL dichloromethane. The organic phases were combined, dried with 214 anhydrous sodium sulfate, and then analyzed by GC-MS, under the conditions described 215 elsewhere.(29) Quantification was performed with calibration curves obtained with red wines. 216

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218 Statistical Analyses

Volatile compound concentrations and oenological parameters (milligrams or micrograms per liter) were expressed as mean value \pm standard deviation. The effects of yeast/LAB combinations were tested using one-way and two-way analysis of variance. Principal component analysis (PCA) was also carried out on the concentrations quantified for certain compounds. Statistical analyses were performed using XL-STAT (Addinsoft, Paris, France), whereas graphical representations of PCA were obtained using R v2.15.0 (R Development Core Team 2009, Vienna, Austria; R Foundation for Statistical Computing).

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227 RESULTS AND DISCUSSION

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229 Fermentation conditions and chemical composition of wines

Six combinations of yeast/LAB starter cultures (three yeasts, two bacteria) were tested in Cabernet Sauvignon wines made under micro-vinification conditions (2 hL). The whole winemaking process, including AF and MLF, highlighting the kinetic performance of the microorganisms, is presented in Fig. 1. Since no significant difference was observed between the triplicate experiments, one representative fermentation curve is presented for each modality.

As shown in Fig. 1a, all AF followed the same pattern and were completed in 7 days 235 (170 h). Total reducing sugar in the must was around 218 g L⁻¹ and no differences were found 236 between wines after AF (<1 g L⁻¹). There was a negligible difference in the ethanol 237 concentrations of the wines, with an average of 13.2% (v/v). The pH value of the must was 238 3.48, which had increased slightly after AF (around 3.51). No significant differences in total or 239 volatile acidity were found between wines. Finally, concentrations of L-malic acid in musts 240 fermented with the 522D and XR yeast strains decreased during AF (0.29 g L⁻¹ and 0.18 g L⁻¹ 241 respectively). This suggested that these two strains had the ability to metabolize malic acid in 242 the presence of glucose or other assimilable carbon sources.(31) 243

After alcoholic fermentation was completed, LAB were inoculated. As shown in Figure 244 1b, MLF was completed in every case, irrespective of the bacteria strain used. However, the 245 degradation kinetics of L-malic acid during the course of MLF varied depending on the LAB 246 strain. All O. oeni 450 samples completed MLF in 26 days (L-malic acid < 0.1 g L⁻¹), 247 irrespective of the yeast strain. In contrast, all B28 samples required much longer to complete 248 MLF: 31 days for XR/B28 and 522D/B28 and 33 days for FX10/B28. It is important to note 249 250 that dissimilarities in the kinetics of these two bacteria strains were not due to a difference in the L-malic acid degradation rate. The latency phase of B28 strain was longer than that of the 251 450 (5 days), suggesting a differential adaptation to growth in wine.(32) 252

253 After 3 months' aging, differences between most of the oenological parameters of the various modalities were negligible. Only volatile acidity, expressed in grams of acetic acid per 254 liter, was significantly affected by the LAB cultures. The largest increase was measured in 255 522D/B28 (0.23 g L⁻¹), XR/B28 (0.29 g L⁻¹), and FX10/B28 (0.30 g L⁻¹) samples, with 256 statistically significant differences depending on the LAB strain used. The influence of bacterial 257 strains on volatile acidity has already been reported.(33,34) Acetic acid is produced from citric 258 acid by some genera of LAB,(10) and the statistically significant differences in acetic acid 259 content observed may be due to degradation of larger quantities of citric acid by O. oeni B28. 260

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262 Influence of yeast/LAB combination on wine aromatic compounds

Seventy-three major volatile compounds were quantified, including eight acids, six alcohols, six aldehydes and ketones, six lactones, four C₁₃-norisoprenoids, two sulfur-containing compounds, one terpene, and 40 esters, using analytical methods that previously developed and validated in our laboratory. Concentrations measured in the different modalities are presented in Tables 2 and 3. First, a one-way analysis of variance (ANOVA) was used to study the yeast/LAB combination parameter. Results revealed a significant effect of the microorganism 269 combination on the concentrations of 51 volatile compounds, mainly alcohols, acids and esters. 270 Varietal compounds and α -dicarbonyl compounds were less affected. Concentrations of 271 aldehydes (glyoxal, methylglyoxal) and volatile sulfur compounds (DMS, H₂S) did not vary 272 according to the microorganism combination.

Larger quantities of higher alcohols are present in alcoholic beverages than any other 273 group of aroma compounds. Their concentrations were significantly affected by the 274 275 yeast/bacteria combination used in winemaking. The total amount of higher alcohols was strongly associated with the concentration of 3-methylbutan-1-ol, which constituted over 60% 276 of the total alcohol for each modality. However, this was the only higher alcohol not affected 277 278 by the yeast/LAB combination, while concentrations of other alcohols, such as propan-1-ol, 2methypropan-1-ol and 2-methylbutan-1-ol differed significantly differences in their according 279 to the yeast/LAB combination (0.1%, 5% and 0.1%, respectively). 280

Eleven varietal compounds known to contribute to the fruity aroma of red wines, including C₁₃-norisoprenoids, lactones and terpene, were quantified. For C₁₃-norisoprenoids, differences between the six combinations were low or non-existent and only α -ionone presented small, but significant, variations (1%). Lactones were mainly represented by γ -octalactone, with significant variations in concentration (5%) according to the yeast/LAB combination. The concentrations of other lactones were not significantly affected by the microorganism combinations.

Eight volatile acids known to contribute to the balance of fruity aroma were assayed. Concentrations of branched acids (isobutyric, isovaleric and 2-methylbutyric acids) were significantly modulated by the yeast/LAB combination (0.1%). Similarly, levels of linear acids (butyric, hexanoic, decanoic and dodecanoic) all varied depending on the microorganism combinations (0.1%, except for decanoic acid, significant at 5%). Finally, esters are considered one of the most important families of aromatic compounds for modulating red wine fruity aromas. Among the 40 esters quantified, only seven were not affected by the yeast/LAB combination. Concentrations of over half of the compounds (33 esters) differed significantly according to the microorganisms used.

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298 Predominant impact of yeast on concentrations of aromatic compounds

Principal component analysis (PCA) was used to refine these observations. Among 73 299 molecules quantified, 22 did not exhibit any significant combination effect and were not 300 included in the PCA. Using 51 analytical variables (volatile data) and 18 objects (3 yeasts x 2 301 302 bacteria in triplicate), PCA explained over 65% of the total variance on the first two axes (Fig. 2). Triplicates of each modality were all represented close to each other, indicating good 303 reproducibility of the experiment. According to this PCA, the yeast strain alone had a greater 304 305 impact on volatile compound levels than the yeast/LAB combination. Indeed, triplicate samples fermented with FX10/450 and FX10/B28 were separated from the other wines along axis 1. 306 307 Samples inoculated with XR/450 and XR/B28 combinations were at the bottom of the twodimensional plot, whereas the 522D/B28 and 522D/450 samples were higher on axis 2. Ethyl 308 lactate and diacetyl were the only compounds strongly represented on axis 3 (10.37% of total 309 310 variance; data not shown), which separated the wines according to the LAB strain, as expected. In contrast, no yeast/LAB combination effect was revealed. 311

These observations were confirmed with a two-way ANOVA (yeast/bacteria/yeast x bacteria interaction) (Table 4). Among the fifty-one compounds previously highlighted, only eight were actually affected by the yeast/LAB interaction, while a yeast strain effect was observed for 48 of these aromatic compounds.

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The concentration of higher alcohols was only modulated by the yeast strain. Wines fermented by the 522D strain contained significantly more 2-methylpropan-1-ol, propan-1-ol, and 2-methylbutan-1-ol than FX10 or XR wines (1%, 0.1% and 0.1%, respectively).

Yeast strains also influenced the C13-norisoprenoid and lactone concentrations, but their 319 impact was not as clear (Table 4). Indeed, only small variations were measured in lactone 320 concentrations. Among these compounds, y-octalactone was the most representative, with 321 levels ranging from 15.86 µg L⁻¹ (XR/B28 wine) to 22.50 µg L⁻¹ (522D/B28 wine), but is 322 unlikely to have had any aromatic impact in view of its perception threshold (35 μ g L⁻¹).(35) 323 Little information is available concerning lactone formation pathways in wine, but they are 324 325 assumed to be mainly synthesized from hydroxylated fatty acids or esters via an enzymatic or chemical pathway.(18,36) The results of this study were consistent with previous observations 326 327 that yeasts were capable of enzymatic esterification but not, apparently, LAB(37). However, 328 lactones are mainly synthesized during wine aging(38) and some differences in concentrations may occur depending on the LAB strain used during MLF. Indeed, some studies have indicated 329 the possibility of a late synthesis of these compounds, related to bacterial β -glycosidase and 330 oxidase activities. (36,39) Among the C₁₃-norisoprenoids, only α -ionone presented small 331 variations in concentration with different yeast or LAB strains, as well as yeast/LAB 332 interactions (from 0.11 µg L⁻¹ for FX10/450 to 0.22 µg L⁻¹ for XR/B28). Although levels found 333 in this study were below the perception threshold (2.6 μ g L⁻¹),(40) which is highly dependent 334 on the matrix, some studies have highlighted the potential implication of these compounds in 335 modulating fruity aroma via perceptive interactions.(17) These results are in accordance with 336 numerous data presented in the literature, demonstrating the ability of both yeast and LAB to 337 hydrolyze glycosidic precursors of C₁₃-norisoprenoids.(36,41) 338

Two groups were identified among the 40 esters quantified in this study. Major esters, including ethyl acetate, ethyl lactate, and monoethyl succinate, were present at higher concentrations (mg L⁻¹) compared to other esters, which are nevertheless considered "odorant
esters", due to their lower perception threshold in wine.

In the major ester group, diethyl succinate and ethyl acetate concentrations were slightly impacted by yeast strains. However, in view of its perception threshold (154 mg L⁻¹)(40) and the variations measured in this study (<20 mg L⁻¹), ethyl acetate probably did not affect wine aroma. Ethyl lactate levels varied significantly among the different modalities, reaching higher concentrations in wines inoculated with LAB strain B28 (0.1%), confirming the literature reporting the capacity of LAB to synthesize this compound during MLF.(42,43)

Concentrations of other esters, known as "odorant esters", were also mainly influenced by the 349 350 yeast strain. Three groups may be identified in terms of their contribution to fruity aroma. Fatty acid ethyl esters were the least influenced by the yeast/LAB combination. Ethyl butyrate, 351 decanoate and dodecanoate, as well as their corresponding acids, were mainly synthesized by 352 353 FX10 and 522D yeasts. Higher concentrations of most acetates were found in wines fermented with 522D and FX10 (except hexyl acetate). Higher concentrations of branched esters, such as 354 ethyl 2-methylbutyrate and ethyl isovalerate, were found in wines fermented with 522D or XR 355 (significant at 0.1%). Similarly, significantly higher levels of the corresponding acids, such as 356 isobutyric, isovaleric and 2-methylbutyric acids, were also found in these last two wines (also 357 358 at 0.1%). The concentrations of the other esters (esters with an odd number of carbon atoms, methyl esters, isoamyl esters, cinnamates and minor esters) were also affected by the different 359 combinations used, particularly the yeast strain, as reported in previous studies.(9,44) Although 360 the variations measured for these esters were below the perception threshold, some studies have 361 demonstrated that they may still be perceived by a trained panelist.(15,45) 362

While the majority of these compounds were synthesized by yeast during AF, the esterase activity of wine LAB has also been reported.(46) Besides diacetyl, known to be synthesized during MFL by LAB,(24) these results suggest that microorganisms may be capable of

modulating the concentrations of esters and their corresponding acids. The carbon chain length 366 seemed to be an important parameter in the synthesis of these compounds by LAB. Indeed, the 367 longer the carbon chain, the more the esters and acids were affected by the LAB strain (Table 368 4). Hexanoic, octanoic, decanoic and dodecanoic acids were all found in significantly higher 369 concentrations in wines inoculated with LAB strain 450 (1%). This was also true of the 370 corresponding esters, ethyl decanoate and ethyl dodecanoate (significant at 0.1%). These results 371 372 contradicted some data in the literature. Matthews *et al.*(47) reported that the hydrolytic activity of esterases in different species or genera (O. oeni, Lactobacillus, Pediococcus) had greater 373 specificity for substrates with short carbon chains (C2, C4). In particular, the esterase activity 374 375 of O. oeni was reported to be greater for substrates in C4. In contrast, other recent studies reported the ability of LAB to synthesize ethyl esters and acetates with long carbon chains (C8, 376 C10, C12).(29) In all cases, these long carbon chain esters play a minor role in red wine fruity 377 378 aroma. It is therefore unlikely that the small variations in concentration observed between samples inoculated with different LAB strains would be perceived by a tasting panel. 379

This study examined the influence of S. cerevisiae and O. oeni strains on the production 380 of Bordeaux red wines using six different yeast/LAB combinations. Results obtained for 381 standard chemical parameters revealed that the level of volatile acidity varied significantly 382 383 according to the LAB strain. For aromatic compounds, each microorganism combination resulted in a specific volatile profile. However, the yeast strain was apparently the predominant 384 component in the yeast/LAB combination in modulating aromatic compound levels. In 385 386 particular, the 522D and FX10 strains exhibited a similar capacity to produce esters, acids and higher alcohols. These results showed that yeasts had a more significant effect on wine quality 387 and are thus likely to have a greater impact on wine style than the LAB used. A previous study 388 had already demonstrated the predominant impact of yeast strain rather than yeast/LAB 389 combination on cherry wines.(48) 390

Sensory analyses were performed on these six wines and presented in a previous 391 392 study,(49) using a Napping® test. According to Napping® results obtained with wines at two different aging steps (3 and 12 months), the differences observed between modalities seemed 393 to be correlated with the yeast strain use for AF. Most descriptors used to discriminate wines 394 referred to fruity notes. In both cases, the trained panel composed of 20 judges perceived FX10 395 and XR wines as being fruitier than 522D wines. To confirm these preliminary results, a ranking 396 397 test and a comparison profile were performed with wines from the 2012 vintage fermented with the same yeast/LAB combinations. In this study,(49) the yeast strain appeared to be a dominant 398 factor involved in the modulation of fruity notes in Bordeaux red wines. Wines inoculated with 399 400 FX10 were perceived as fruitier, regardless of the vintage or grape cultivar, after 3 and 12 months of aging. 401

If we consider the volatile composition of these wines, samples fermented with the yeast FX10 402 403 had higher values for the attributes referring to "fruity", due to their large quantities of fruity ethyl esters. Surprisingly, 522D wines, described as fruitless, also contained important levels 404 405 of these aromatic compounds, as well as high amounts of higher alcohols. These compounds, recognized by their strong, pungent smell, influence the taste and character of wine depending 406 on their concentration: below 300 mg L⁻¹, they contribute to the desirable complexity of wine 407 but at concentrations exceeding 400 mg L⁻¹, they are regarded as a negative influence on wine 408 quality.(21) The high alcohol levels found in this study, particularly in 522D/B28 and 522D/450 409 samples (577 mg L⁻¹ and 570 mg L⁻¹, respectively), may have had a negative effect on fruity 410 aroma perception in these wines. 411

While these experiments offer new insights into the organoleptic effect of fermentations, the chemistry underlying the sensory interactions is highly complex. Further investigations are necessary to elucidate the influence of yeast- and LAB-derivative compounds on fruity aroma. Moreover, in light of recent articles dealing with the interactions between volatile and non-

- volatile compounds,(50) the impact of both microorganisms on the non-volatile matrix should
- 417 also be investigated as a potential modulating factor of wine aroma.

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TABLES

	XR/B28 [*]	522D/B28	FX10/B28	XR/450	522D/450	FX10/450
Fermentation duration (days)	47	47	47	40	42	35
Alcoholic degree (% v/v)	13.2 ± 0.2	13.2 ± 0.3	13.2 ± 0.2	13.2 ± 0.1	13.2 ± 0.3	13.1 ± 0.2
рН	3.68 ± 0.03	3.67 ± 0.01	3.60 ± 0.02	3.64 ± 0.03	3.64 ± 0.01	3.59 ± 0.02
Total acidity (g L^{-1} H_2SO_4)	3.4 ± 0.04 cd	3.43 ± 0.08 cd	3.56 ± 0.03 b	3.5 ± 0.1 c	3.54 ± 0.04 bc	3.67 ± 0.02 a
Volatile acidity (g L ⁻¹ acetic acid)	0.29 ± 0.02 a	0.23 ± 0.03 ab	0.30 ± 0.02 a	0.13 ± 0.02 c	0.09 ± 0.01 d	0.16 ± 0.01 c
Total sulfur dioxide (mg L ⁻¹)	41 ± 2	43 ± 6	42 ± 4	33 ± 9	39 ± 13	29 ± 1
Free sulfur dioxide (mg L ⁻¹)	30 ± 4	28 ± 2	30 ± 2	20 ± 5	25 ± 7	21 ± 1

Table 1. Mean Concentration with Standard Deviation of Oenological Parameters of Wines after 3 months' aging.

 * Values with different superscript roman letter (a-d) in the same row are significantly different according to the Tuckey *post hoc* test (P < 0.05).

Compounds	XR/B28	XR/450	522D/B28	552D/450	FX10/B28	FX10/450	One-way ANOVA
Alcohols							
propan-1-ol	35 ± 5	37 ± 3	55 ± 5	55 ± 5	38 ± 2	40 ± 2	***
2-methylpropan-1-ol	54 ± 2	56 ± 2	61 ± 3	62 ± 3	57 ± 1	58.4 ± 0.4	*
2-methylbutan-1-ol	88 ± 5	90 ± 4	115 ± 4	114 ± 8	79 ± 1	80 ± 3	***
3-methylbutan-1-ol	$319~\pm~17$	$330~\pm~10$	$346~\pm~13$	$341~\pm~21$	326 ± 8	334 ± 7	NS
Sum higher alcohols	496	514	577	570	499	513	
butan-2,3-diol (D)	127 ± 31	99 ± 16	140 ± 17	98 ± 16	189 ± 66	116 ± 9	*
butan-2,3-diol (M)	$49~\pm~17$	56 ± 9	54 ± 8	52 ± 11	66 ± 21	52 ± 4	NS
Aldehydes & ketones							
glyoxal	$0.14 ~\pm~ 0.03$	$0.1~\pm~0.01$	$0.16~\pm~0.05$	$0.19 ~\pm~ 0.05$	$0.15~\pm~0.02$	$0.13 ~\pm~ 0.03$	NS
methylglyoxal	$0.4~\pm~0.06$	$0.42 \ \pm \ 0.07$	$0.46~\pm~0.06$	$0.45~\pm~0.06$	$0.55~\pm~0.08$	$0.43 \ \pm \ 0.03$	NS
acetoin	19 ± 4	24 ± 3	24 ± 5	19 ± 2	30 ± 10	21 ± 2	NS
diacetyl	11 ± 1	7.4 ± 0.6	10 ± 1	7 ± 2	10 ± 1	5.6 ± 0.5	***
pentan-2,3-dione	1.5 ± 0.1	1.68 ± 0.09	1.47 ± 0.01	1.9 ± 0.2	1.1 ± 0.2	1.4 ± 0.2	***
frambinone (μ g/L)	15 ± 5	14 ± 3	14 ± 3	13 ± 3	12 ± 2	$11.1~\pm~0.7$	NS
Sulfur-containing com	ipounds						
hydrogen sulfide	$0.8~\pm~0.1$	$0.7~\pm~0.2$	0.9 ± 0.3	$1.0~\pm~0.4$	1.2 ± 0.6	1.2 ± 0.2	NS
dimethyl sulfide	$3.7~\pm~0.2$	$3.7~\pm~0.2$	$4.0~\pm~0.5$	3.6 ± 0.7	$4.1~\pm~0.3$	$4.0~\pm~0.4$	NS
Acids							
butyric acid	7.5 ± 1.0	5.5 ± 0.5	6.7 ± 0.6	$4.6~\pm~0.4$	5.2 ± 0.3	3.7 ± 0.2	***
isobutyric acid	1.22 ± 0.09	1.2 ± 0.1	1.22 ± 0.03	$1.22 ~\pm~ 0.03$	1.01 ± 0.01	$1.00~\pm~0.02$	***
isovaleric acid	$1.11 ~\pm~ 0.07$	$1.1~\pm~0.07$	1.11 ± 0.04	1.12 ± 0.01	$0.76~\pm~0.01$	$0.78~\pm~0.02$	***

Table 2. Mean concentrations with standard deviation (mg L^{-1} , n=3) of fermentation-derived compounds in wines made by different yeast/LAB combinations.

2-methylbutyric acid	$0.82~\pm~0.04$	$0.8~\pm~0.05$	$0.94~\pm~0.06$	$0.93~\pm~0.03$	$0.51 ~\pm~ 0.02$	$0.51~\pm~0.01$	***
hexanoic acid	7.8 ± 0.4	$8.0~\pm~0.4$	$8.7~\pm~0.4$	$9.4~\pm~0.3$	$8.4~\pm~0.5$	9.2 ± 0.5	***
octanoic acid	$2.6~\pm~0.1$	$3.0~\pm~0.3$	$2.9~\pm~0.1$	$3.4~\pm~0.2$	3.3 ± 0.2	$3.62~\pm~0.05$	***
decanoic acid	$0.7 ~\pm~ 0.04$	$0.76~\pm~0.06$	$0.75~\pm~0.05$	$0.88 ~\pm~ 0.09$	$0.72 \ \pm \ 0.04$	$0.82~\pm~0.07$	*
dodecanoic acid (µg/L)	9 ± 3	6 ± 1	$6.7 ~\pm~ 0.8$	11 ± 1	$6.8~\pm~0.7$	$12.1~\pm~0.8$	***
C ₁₃ -norisoprenoids, lact	ones, & terpene						
β -damascone (μ g/L)	$0.03~\pm~0.01$	$0.03 ~\pm~ 0.00$	$0.03~\pm~0.01$	$0.04~\pm~0.01$	$0.05~\pm~0.04$	$0.02 ~\pm~ 0.01$	NS
β -damascenone (μ g/L)	$6.6~\pm~0.2$	$6.0~\pm~0.4$	5.5 ± 0.6	$6.3~\pm~0.9$	$6.1~\pm~0.5$	$6.5~\pm~0.5$	NS
α -ionone (μ g/L)	$0.22 ~\pm~ 0.02$	$0.12 ~\pm~ 0.02$	$0.15~\pm~0.02$	$0.12 \ \pm \ 0.03$	$0.14~\pm~0.04$	$0.11 ~\pm~ 0.02$	**
β -ionone (μ g/L)	$0.09~\pm~0.02$	$0.09~\pm~0.01$	$0.08~\pm~0.01$	$0.1~\pm~0.01$	$0.1~\pm~0.02$	$0.1~\pm~0.01$	NS
γ -octalactone (μ g/L)	16 ± 3	17 ± 3	23 ± 2	21 ± 2	19 ± 3	18 ± 2	*
γ -nonalactone (μ g/L)	$7.7~\pm~0.4$	$7.1~\pm~0.5$	8 ± 1	8 ± 1	8 ± 2	7.1 ± 0.5	NS
γ -decalactone (μ g/L)	$0.91 ~\pm~ 0.01$	$0.8~\pm~0.2$	$0.63~\pm~0.08$	$0.62~\pm~0.06$	$0.8~\pm~0.3$	$0.7~\pm~0.01$	NS
δ -decalactone (μ g/L)	$1.06~\pm~0.04$	$1.3~\pm~0.3$	1.2 ± 0.1	$1.3~\pm~0.1$	$1.9~\pm~0.5$	$1.57~\pm~0.09$	*
γ -undecalactone (µg/L)	$0.07 ~\pm~ 0.01$	$0.09~\pm~0.02$	$0.08~\pm~0.01$	$0.07 ~\pm~ 0.01$	$0.09~\pm~0.01$	$0.08~\pm~0.01$	NS
γ -dodecalactone (μ g/L)	$0.05~\pm~0.01$	$0.05~\pm~0.02$	$0.06~\pm~0.01$	$0.05~\pm~0.01$	$0.06~\pm~0.01$	$0.05~\pm~0.01$	NS
linalol (µg/L)	12 ± 2	$9.7~\pm~0.7$	13 ± 2	12 ± 1	8 ± 2	$8.0~\pm~0.4$	**

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Compounds	XR/B28	XR/450	522D/B28	552D/450	FX10/B28	FX10/450	One-way ANOVA
Major polar esters							
ethyl lactate (mg/L)	56 ± 5	31 ± 1	55 ± 2	31 ± 2	55 ± 1	38 ± 4	***
monoethyl succinate							
(mg/L)	22 ± 1	20 ± 2	22 ± 3	20 ± 1	21 ± 2	20 ± 1	NS
diethyl succinate	683 ± 29	586 ± 38	$793~\pm~105$	697 ± 115	621 ± 81	$588~\pm~50$	*
Polar esters							
ethyl leucate	70 ± 5	$58.2~\pm~0.3$	94 ± 6	96 ± 13	85 ± 13	$80~\pm~16$	**
ethyl 3-hydroxybutyrate	$333~\pm~10$	$323~\pm~12$	$384~\pm~29$	$387~\pm~18$	$454~\pm~26$	$454~\pm~21$	***
ethyl 2-hydroxyhexanoate	$0.9~\pm~0.3$	1.6 ± 0.6	$1.27 ~\pm~ 0.04$	1.2 ± 0.2	$1.2~\pm~0.2$	$1.0~\pm~0.1$	NS
ethyl 6-hydroxyhexanoate	$3.12 ~\pm~ 0.06$	$3.4~\pm~0.8$	$3.1~\pm~0.7$	4 ± 1	$3.9~\pm~0.4$	$3.9~\pm~0.7$	NS
Ethyl fatty acid esters							
ethyl butyrate	185 ± 17	179 ± 8	$218~\pm~22$	$218~\pm~28$	198 ± 9	194 ± 17	*
ethyl hexanoate	$286~\pm~11$	$294~\pm~5$	$319~\pm~13$	$320~\pm~23$	$313~\pm~18$	$324~\pm~29$	NS
ethyl octanoate	$289~\pm~20$	$282~\pm~25$	$307~\pm~27$	$302~\pm~34$	330 ± 21	334 ± 15	NS
ethyl decanoate	71 ± 6	94 ± 3	91 ± 10	103 ± 8	87 ± 6	115 ± 7	***
ethyl dodecanoate	$4.8~\pm~0.4$	$6.8~\pm~0.4$	$7.4~\pm~0.6$	$9.9~\pm~0.9$	8 ± 1	13 ± 2	***
Ethyl branched acid esters							
ethyl isobutyrate	62 ± 6	58 ± 6	63.5 ± 0.9	$49.0~\pm~15.0$	61 ± 4	59 ± 6	NS
ethyl 2-methylbutyrate	11.7 ± 0.7	12 ± 1	13.7 ± 0.3	$13.8~\pm~0.8$	7.4 ± 0.2	7.1 ± 0.3	***
ethyl isovalerate	18.4 ± 0.4	17 ± 1	18 ± 1	19 ± 1	11.6 ± 0.4	11.7 ± 0.6	***
ethyl phenylacetate	$3.22~\pm~0.02$	3.2 ± 0.4	$5.0~\pm~0.6$	5.5 ± 0.5	$3.0~\pm~0.3$	$3.1~\pm~0.1\mu$	***
Acetate of higher alcohols							
ethyl acetate (mg/L)	90 ± 2	78 ± 3	72 ± 1	71 ± 4	88 ± 4	80 ± 3	***
propyl acetate	17.1 ± 0.5	$15.5~\pm~0.5$	23 ± 2	21 ± 1	21 ± 2	$19.3~\pm~0.3$	***

Table 3. Mean concentrations with standard deviation ($\mu g L^{-1}$, n=3) of ester compounds in wines made with different yeast/LAB combinations.

isobutyl acetate	41 ± 1	37 ± 2	53 ± 4	51 ± 5	50 ± 2	51 ± 1	***
butyl acetate	$0.8~\pm~0.2$	$0.8~\pm~0.1$	$0.8~\pm~0.1$	1.2 ± 0.2	1.26 ± 0.07	$0.8~\pm~0.2$	**
isoamyl acetate	$1105~\pm~61$	$1064 \ \pm \ 37$	$1484~\pm~181$	$1339~\pm~196$	$1367~\pm~143$	$1317~\pm~73$	**
hexyl acetate	6 ± 1	9 ± 2	7 ± 1	9 ± 2	$8.1~\pm~0.9$	$8.7~\pm~0.3$	NS
octyl acetate	$0.08~\pm~0.01$	$0.12 ~\pm~ 0.02$	$0.1~\pm~0.01$	$0.15~\pm~0.04$	$0.15~\pm~0.03$	$0.18~\pm~0.01$	***
2-phenylethyl acetate	$87~\pm~8$	$89~\pm~3$	$144~\pm~24$	$145~\pm~17$	$117~\pm~10$	$120~\pm~9$	***
Methyl esters							
methyl butyrate	$0.86~\pm~0.05$	$0.78~\pm~0.02$	1.22 ± 0.01	$1.09~\pm~0.08$	1.0 ± 0.1	$0.84~\pm~0.04$	***
methyl hexanoate	1.9 ± 0.1	1.71 ± 0.06	2.2 ± 0.2	$2.16~\pm~0.09$	1.9 ± 0.2	2.0 ± 0.1	**
methyl octanoate	1.26 ± 0.02	$1.34~\pm~0.05$	1.4 ± 0.1	1.5 ± 0.1	$1.44 ~\pm~ 0.09$	1.5 ± 0.1	**
methyl decanoate	$0.33~\pm~0.01$	$0.42 ~\pm~ 0.03$	$0.42 ~\pm~ 0.03$	$0.48~\pm~0.03$	$0.4~\pm~0.01$	$0.48~\pm~0.01$	***
Ethyl esters with odd nur	nber of carbon ato	ms					
ethyl propanoate	306 ± 12	$292~\pm~8$	$425~\pm~25$	384 ± 58	$281~\pm~22$	$258~\pm~16$	***
ethyl valerate	$0.67~\pm~0.03$	$0.54~\pm~0.04$	$1.0~\pm~0.1$	$0.87 ~\pm~ 0.03$	$0.84~\pm~0.06$	$0.9~\pm~0.2$	***
ethyl heptanoate	$0.9~\pm~0.1$	$0.9~\pm~0.1$	$0.92~\pm~0.06$	$0.95~\pm~0.06$	0.64 ± 0.01	$0.66~\pm~0.03$	***
ethyl nonanoate	$0.61~\pm~0.01$	$0.69~\pm~0.03$	$0.89~\pm~0.03$	$0.9~\pm~0.04$	$0.89~\pm~0.08$	$1.07~\pm~0.09$	***
Isoamyl esters							
isoamyl butyrate	$0.66~\pm~0.06$	$0.67 ~\pm~ 0.05$	$0.75~\pm~0.02$	$0.8~\pm~0.1$	$0.71 ~\pm~ 0.05$	$0.57~\pm~0.03$	***
isoamyl hexanoate	1.9 ± 0.1	1.88 ± 0.06	$2.0~\pm~0.1$	1.9 ± 0.2	2.1 ± 0.1	$2.18~\pm~0.07$	*
isoamyl octanoate	$2.8~\pm~0.3$	$2.9~\pm~0.2$	3.1 ± 0.3	$3.2~\pm~0.2$	3.4 ± 0.1	$3.48~\pm~0.07$	**
Cinnamates							
ethyl cinnamate	$2.55~\pm~0.02$	$2.57~\pm~0.05$	2.4 ± 0.1	2.4 ± 0.2	$2.56~\pm~0.07$	$2.71 ~\pm~ 0.06$	**
ethyl dihydrocinnamate	1.7 ± 0.1	$1.69~\pm~0.05$	1.5 ± 0.1	$1.52~\pm~0.06$	$1.51~\pm~0.07$	$1.57~\pm~0.05$	**
Minor esters							
ethyl hexenoate	1.8 ± 0.2	1.6 ± 0.1	$1.41 ~\pm~ 0.07$	$1.51~\pm~0.09$	1.3 ± 0.1	$1.68~\pm~0.09$	***
isobutyl hexanoate	$0.16~\pm~0.00$	$0.17 ~\pm~ 0.01$	$0.16~\pm~0.01$	$0.17 ~\pm~ 0.01$	$0.2~\pm~0.01$	$0.21~\pm~0.01$	***

Significant effect: NS, Not Significant; * P < 0.05; ** P < 0.01; *** P < 0.001

Compounds	PCA	two-way ANOVA			
Compounds	abbreviation	yeast	bacteria	yeast x bacteria	
Alcohols					
propan-1-ol	СЗОН	***	NS	NS	
2-methylpropan-1-ol	2mC3OH	**	NS	NS	
2-methylbutan-1-ol	2mC4OH	***	NS	NS	
butan-2,3-diol (D)	С4-2,3ОН	NS	**	NS	
Aldehydes & ketones					
diacetyl	diacetyl	NS	***	NS	
pentan-2,3-dione	pentan-2,3-dione	***	***	NS	
Acids					
butyric acid	C4	***	***	NS	
isobutyric acid	iC4	***	NS	NS	
isovaleric acid	iC5	***	NS	NS	
2-methylbutyric acid	2mC4	***	NS	NS	
hexanoic acid	C6	***	**	NS	
octanoic acid	C8	***	***	NS	
decanoic acid	C10	*	**	NS	
dodecanoic acid	C12	*	**	***	
C13-norisoprenoids, lacto	ones, & terpene				
α-ionone	α-i	*	***	*	
γ-octalactone	γ-oct	**	NS	NS	
δ-decalactone	δ-dec	**	NS	NS	
linalol	linalol	**	NS	NS	
Major polar esters					
ethyl lactate	C2lac	NS	***	NS	
diethyl succinate	DES	*	NS	NS	
Polar esters					
ethyl leucate	C2leu	**	NS	NS	
ethyl 3-hydroxybutyrate	C2 30HC2	***	NS	NS	
Ethyl fatty acid esters					
ethyl butyrate	C2C4	**	NS	NS	
ethyl decanoate	C2C10	**	***	NS	
ethyl dodecanoate	C2C12	***	***	*	

Table 4. Results of the two-way ANOVA (yeast/LAB/yeast x LAB interaction).

Table 4. Continued

Compounds	PCA		two-way ANOVA			
Compounds	abbreviation	yeast	bacteria	yeast x bacteria		
Ethyl branched acid ester	\$					
ethyl 2-methylbutyrate	C2 2mC4	***	NS	NS		
ethyl isovalerate	C2iC5	***	NS	NS		
ethyl phenylacetate	C2PhC2	***	NS	NS		
Acetate of higher alcohol	5					
ethyl acetate	C2C2	***	***	**		
propyl acetate	C3C2	***	**	NS		
sobutyl acetate	iC4C2	***	NS	NS		
butyl acetate	C4C2	*	NS	**		
soamyl acetate	iC5C2	**	NS	NS		
octyl acetate	C8C2	***	**	NS		
2-phenylethyl acetate	2-PhC2C2	***	NS	NS		
Methyl esters						
methyl butyrate	C1C4	***	**	NS		
nethyl hexanoate	C1C6	***	NS	NS		
nethyl octanoate	C1C8	**	*	NS		
nethyl decanoate	C1C10	***	***	NS		
Ethyl	esters with odd ni	umber of	carbon atom	S		
ethyl propanoate	C2C3	***	*	NS		
ethyl valerate	C2C5	***	NS	NS		
ethyl heptanoate	C2C7	***	NS	NS		
ethyl nonanoate	C2C9	***	**	*		
Isoamyl esters						
isoamyl butyrate	iC5C4	***	NS	**		
soamyl hexanoate	iC5C6	**	NS	NS		
soamyl octanoate	iC5C8	**	NS	NS		
Cinnamates						
ethyl cinnamate	C2cin	**	NS	NS		
ethyl dihydrocinnamate	C2dhicin	***	NS	NS		
Minor esters						
ethyl hexenoate	C2hex	**	*	**		
isobutyl hexanoate	iC4C6	***	NS	NS		
methyl trans-geranate	C1ger	***	***	NS		

Significant effect: NS, Not Significant; * P < 0.05; ** P < 0.01; *** P < 0.001

FIGURE GRAPHICS

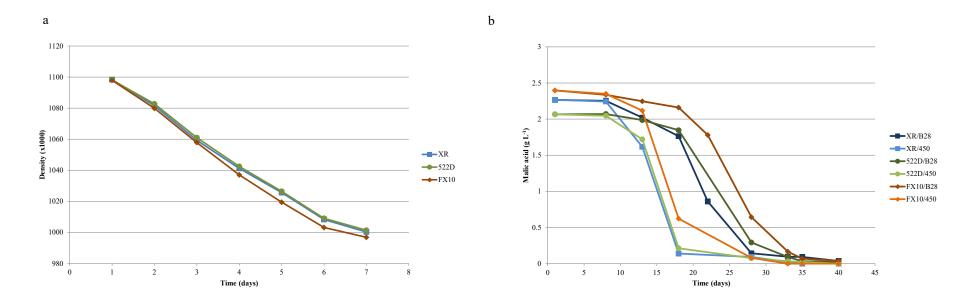


Figure 1. Kinetics of alcoholic (a) and malolactic (b) fermentations in wines fermented with different yeast/LAB combinations.

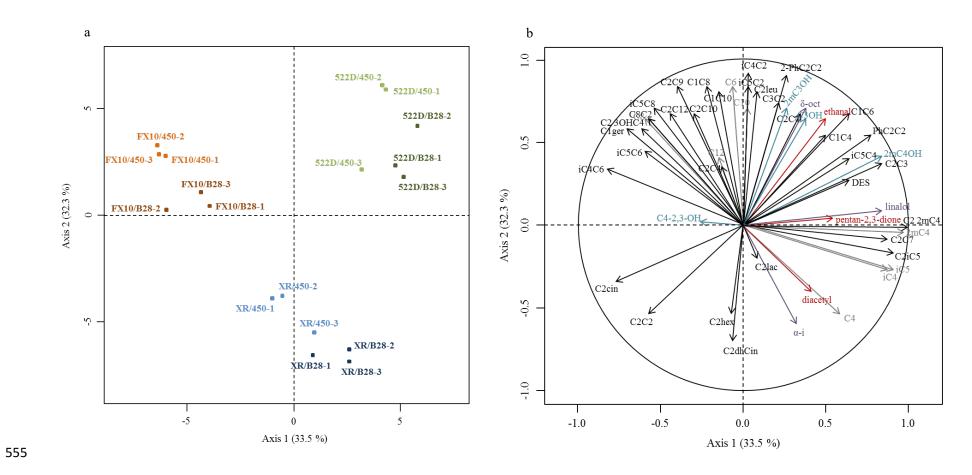


Figure 2. Principal component analysis represented as a scatter point plot (a) and 51 parameters (b) on axes 1 x 2. Abbreviations for the various
 parameters are presented in Table 4.